

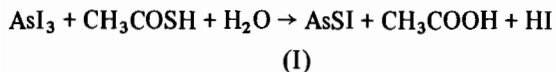
**The Molecular and Crystal Structure of a 1:1-Adduct of AsI<sub>3</sub>, prepared from 1,3,5,7-(tetramethyl)-2,4,6,8,9,10-(hexathia)adamantane**

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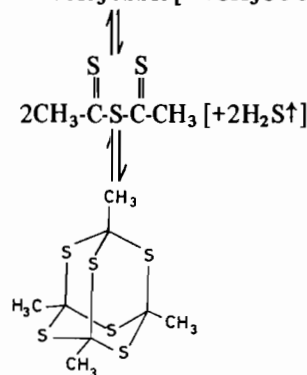
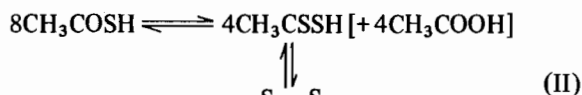
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In order to prepare the ternary sulfoiodide AsSI [1] the reaction of AsI<sub>3</sub> with thioacetic acid [2] was tested:



A mixture of thioacetic acid and powdered AsI<sub>3</sub> was held at room temperature in a closed vessel for four weeks. After a period of 9 days, compact orange-red crystals were grown on top of the AsI<sub>3</sub>-solid; in the course of five more days they were partially transformed to yellow monoclinic plates. Finally a third phase was observed, as white platy crystals. The three different phases were isolated (they can be recrystallized in ether and ethanol) and shown by energy-dispersive X-ray analysis to contain significant amounts of sulfur. Arsenic and iodine were only detected in the orange-red and yellow crystals, the latter containing smaller amounts of these elements. In accordance with this observation and AsI<sub>3</sub>-ion peak was present only in the mass spectra of the orange-red and yellow phases. As confirmed by a complete crystal structure analysis of the orange-red crystals (see below), the compound is a 1:1-adduct of AsI<sub>3</sub> with 1,3,5,7-(tetramethyl)-2,4,6,8,9,10-(hexathia)-adamantane. This led to the positive interpretation of a respective mass-peak as belonging to the tetramethyl-hexathia-adamantane molecular ion; this peak is also present in the mass-spectra of the yellow and the white crystals. With the additional consideration of the melting point (224 °C; DTA-investigation), the white crystals were identified as pure tetramethyl-hexathia-adamantane. The yellow crystals (mp 109 °C) are assumed to be an adduct of AsI<sub>3</sub> with tetramethyl-hexathia-adamantane, but in contrast to the orange-red 1:1-compound with a respective molar ratio < 1.

Formation of tetramethyl-hexathia-adamantane thus probably takes place as follows, and not as suggested in reaction I:



Reaction II is supposed to be catalysed by HI, which is formed simultaneously by reaction of AsI<sub>3</sub> with water. A similar catalytic effect is observed in the presence of HBr [3].

Single crystals of AsI<sub>3</sub>·tetramethyl-hexathia-adamantane were investigated by X-ray methods: AsI<sub>3</sub>·(CH<sub>3</sub>)<sub>4</sub>C<sub>4</sub>S<sub>6</sub>; monoclinic *P*2<sub>1</sub>/*c*; *a* = 1119.9(2), *b* = 1183.9(2), *c* = 1572.4(2) pm, β = 103.06(1)°; *Z* = 4; *D*<sub>obs.</sub> = 2.53, *D*<sub>calc.</sub> = 2.47 g cm<sup>-3</sup>. 3583 independent reflections up to 2θ = 50° (MoKα) were collected on an automated diffractometer (Syntex P2<sub>1</sub>); 2713 reflections with *I* > 1.96σ<sub>*I*</sub> were classified as observed. The phase problem was solved by direct methods (E-XTL). Including an empirical correction for absorption, the anisotropic refinement of the heavy-atom parameters was finished with *R* = 7.9%. Five of the twelve hydrogen atoms were located in a Difference-Fourier-Map; positions of the remaining seven hydrogens were calculated. The final refinement was carried out with fixed parameters of all hydrogen atoms (*R* = 7.8%). A list of the observed and calculated structure factors, as well as anisotropic thermal parameters of the heavy-atoms and fractional coordinates of the hydrogen positions, can be obtained from the authors. The final non-hydrogen atomic parameters are given in Table I.

Figure 1 shows the molecular structure of the adduct of AsI<sub>3</sub> with 1,3,5,7-(tetramethyl)-2,4,6,8,9,10-(hexathia)-adamantane. Bond lengths and bond angles (minimum and maximum values) are given in Table II. The arsenic atom bond to three terminal iodine atoms in a slightly distorted trigonal-pyramidal arrangement with a mean bond length of 257.6 pm; this value is in good agreement with the respective distance in the crystal structure of AsI<sub>3</sub> (259.1 pm, [4]). The coordination-sphere of the arsenic atom is modified to a distorted octahedron by additional contacts to three of the six sulfur atoms of the adamantane molecule. Orientation of the remaining lone-pair electrons of the arsenic atom is assumed to be towards the centre of the six-membered ring.

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TABLE I. Positional Parameters of the Non-Hydrogen Atoms in the Crystal Structure of  $\text{AsI}_3 \cdot 1,3,5,7\text{-}(\text{tetramethyl})\text{-}2,4,6,8,9,10\text{-}(\text{hexathia})\text{-adamantane}$ . Notation of the Atoms is in Accordance with Fig. 1.

| Atom | x          | y          | z          |
|------|------------|------------|------------|
| As   | 0.5834(1)  | 0.2550(1)  | 0.9479(1)  |
| I(1) | 0.5491(1)  | 0.1281(2)  | 0.8116(1)  |
| I(2) | 0.5093(1)  | 0.4442(1)  | 0.8733(1)  |
| I(3) | 0.3925(2)  | 0.1970(2)  | 1.0032(1)  |
| S(1) | 0.8695(3)  | 0.3356(3)  | 0.9708(2)  |
| S(2) | 1.0331(4)  | 0.3923(3)  | 1.1461(3)  |
| S(3) | 0.9870(4)  | 0.1590(4)  | 1.2133(2)  |
| S(4) | 0.8229(3)  | 0.1030(3)  | 1.0376(2)  |
| S(5) | 1.0830(3)  | 0.1837(3)  | 1.0492(2)  |
| S(6) | 0.7733(3)  | 0.3130(3)  | 1.1367(2)  |
| C(1) | 0.9223(11) | 0.1901(11) | 0.9893(9)  |
| C(2) | 0.8750(12) | 0.3870(9)  | 1.0813(9)  |
| C(3) | 1.0763(12) | 0.2437(14) | 1.1565(11) |
| C(4) | 0.8331(12) | 0.1698(11) | 1.1441(8)  |
| C(5) | 0.9229(15) | 0.1432(13) | 0.8980(9)  |
| C(6) | 0.8290(16) | 0.5117(11) | 1.0727(12) |
| C(7) | 1.2112(14) | 0.2415(17) | 1.2095(12) |
| C(8) | 0.7513(15) | 0.0990(15) | 1.1888(11) |

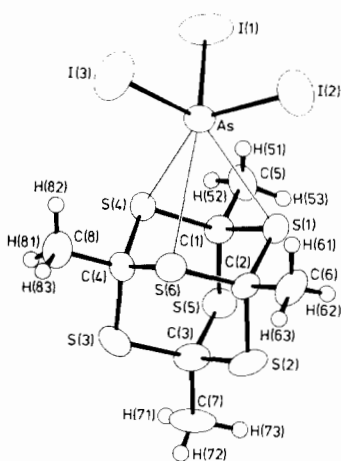


Fig. 1. Molecular structure of the 1:1-adduct of  $\text{AsI}_3$  with 1,3,5,7-(tetramethyl)-2,4,6,8,9,10-(hexathia)-adamantane. Notation of the atoms is in accordance with Table I.

Figure 2 shows the arrangement of the molecular adducts in the crystal structure of  $\text{AsI}_3 \cdot 1,3,5,7\text{-}(\text{tetramethyl})\text{-}2,4,6,8,9,10\text{-}(\text{hexathia})\text{-adamantane}$ , as a layer-structure with alternating layers of  $\text{AsI}_3$  molecules and adamantane molecules parallel to the (100)-plane.

#### Acknowledgements

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TABLE II. Bond Lengths (pm) and Bond Angles ( $^\circ$ ) in  $\text{AsI}_3 \cdot 1,3,5,7\text{-}(\text{tetramethyl})\text{-}2,4,6,8,9,10\text{-}(\text{hexathia})\text{-adamantane}$  (minimum and maximum values). E.s.d.s except for [ ] are 0.2–2.3 pm and 0.1–1.1 $^\circ$ .

| Bond Lengths |             |
|--------------|-------------|
| As–I         | 257.5–257.7 |
| As–S         | 327.4–331.0 |
| S–C          | 179.1–184.7 |
| C–C          | 152.5–155.9 |
| [C–H         | 91–109]     |
| Bond Angles  |             |
| I–As–I       | 98.4–100.3  |
| I–As–S       | 89.5–111.7  |
| S–As–S       | 144.3–154.7 |
| C–S–S        | 55.1–55.2   |
| C–S–C        | 102.5–102.9 |
| S–C–S        | 110.7–114.7 |
| S–C–C        | 104.4–108.6 |
| C–C–H        | 92–116      |
| H–C–H        | 100–123     |

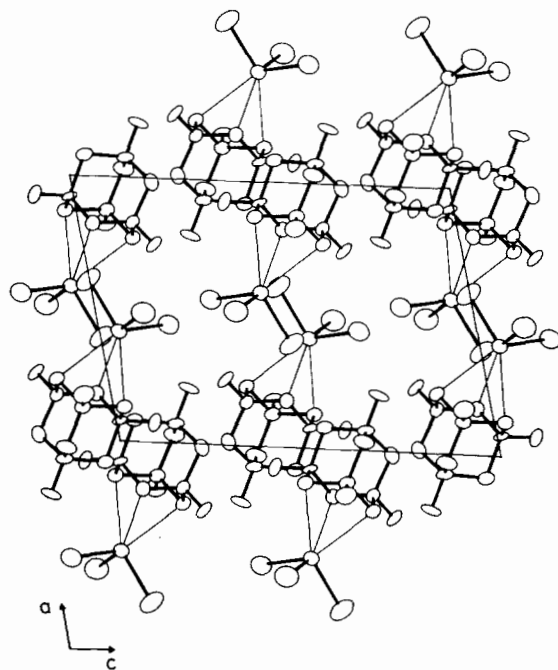


Fig. 2. Arrangement of the molecular adducts in the crystal structure of  $\text{AsI}_3 \cdot 1,3,5,7\text{-}(\text{tetramethyl})\text{-}2,4,6,8,9,10\text{-}(\text{hexathia})\text{-adamantane}$ . Projection along the crystallographic b-axis direction.

#### References

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